

The 250 Most-Cited Primary Authors, 1961-1975. Part III. Each Author's Most-Cited Publication

With Minister 19, 1977

Previously we have listed the 250 most-cited primary authors. We have described how the names were selected.¹ And we have examined the correlation between citedness and other forms of science recognition such as the Nobel prize and membership in national academies of science.² In this essay, the last of three parts, we have listed each author's most-cited publication.

The list appears on pages 11-20. It contains the most-cited publication for which each author was primary author. Textbooks, manuals, reviews, and other items not considered reports of original research were excluded. About half of these publications have appeared in our previous lists of highly-cited items.

The total citation count for each item is based on *Science Citation Index*[®] (*SCI*[®]) data from 1961 to 1975, or from the year of publication if published after 1961. Since one of the 250 most-cited authors was omitted to symbolize the incompleteness of such lists, this list actually contains 249 items.

It is notable that the list contains several items which have been cited relatively few times. Every one of the 250 most-cited authors has been cited over 4,000 times. So some readers may wonder about the paper by R M Barrer, for example, which was cited "only" 46 times. Most of the authors on this list have produced not only high-quality work, but also a large quantity of it. Barrer has published over 300 papers! His most cited publication, a 1959 textbook in physical chemistry,³ had been cited 319 times through 1975. But we have not listed textbooks.

Over four-fifths of these papers were written in the 1950s and 1960s. Four were published in the 1920s, 8 in the 1930s, 29 in the 1940s, 103 in the 1950s, 102 in the 1960s, and 2 in the 1970s. The oldest work on the list is Freud's, which was first published in 1915. The most recent item on the list is Allison's 1971 paper on functions of thymus-derived lymphocytes in relation to autoimmunity.

Thirteen of these 250 papers have been featured in Citation Classics, the weekly *Current Contents*[®] $(CC^{\text{@}})$ series in which authors comment on their classic papers. For each Citation Classic we have indicated the CC issue number, date and page number in brackets after the item. Undoubtedly many more of these papers will appear as Citation Classics in the future.

Just 17 journals account for over half of the articles, and less than a hundred journals account for all the articles. The Journal of the American Chemical Society (JACS) accounts for 26—over 10% of the total. The Journal of Biological Chemistry accounts for 15, Biochemical Journal 11, Physical Review 10, and Journal of Chemical Physics 6. Angewandte Chemie, Biochimica et Biophysica Acta, Journal of Cell Biology, Journal of Molecular Biology and Journal of Physiology account for 4 items each.

Several of the articles are parts of a series. The citation counts include citations only to the part listed. For instance, Seyferth's article is the second of a 79-part series on halomethyl-metal compounds. The last part was published in 1976, after which Seyferth and his colleagues ended their research in this field.

I previously discussed the fact that several scientific fields are not represented by the 250 most-cited primary authors.¹ Since botany was among the unrepresented fields, some readers may be surprised to see that D I Arnon's paper was published in *Plant Physiology*. However, Dr. Arnon assures us that he is *not* a botanist; his field is biochemistry. He explains, "At the time this article was published, there were not many chemists or biochemists interested in chloroplasts. Therefore I didn't publish it in a biochemistry journal. Today I would just as soon put it in *Biochimica et Biophysica Acta*, for example.''⁴

Of course, some of these authors are well known for publications which do not appear on this list. For example, P W Bridgman is well known for his writings on the philosophy of science, which include his books The Way Things Are^5 and The Logic of Modern Physics.⁶

This list contains 40 books and 209 journal articles. Where multiple editions of a book have been published, we counted the citations to *all* editions of the book, but listed the publication date of the earliest edition. However, G W Snedecor's 1937 book on statistical methods was substantially revised and coauthored with W G Cochran in the 1956 edition.⁷ This was the mostcited edition, but the citation count includes citations to all editions.

Collecting citation data on books was a bit tricky for several reasons. First, citation practices concerning books are far from uniform. Second, many of the books which appear on this list are classics which have gone through several editions.

The difficulties can be illustrated by the case of Sigmund Freud. With 8,490 citations, he is still among the most-cited authors, even though he died in 1939. In 1957 the Hogarth Press collected Freud's complete works into a multi-volume Standard Edition. This is often cited simply as "Standard Edition," "S.E.," or

"Standard Ed." Thus, it appears in the SCI in all three forms. In addition, citations to Freud's work sometimes refer to the original publications. After examining the citations to all of Freud's works, it became apparent that Volume 14, concerning the history of the psychoanalytic movement, is the mostcited volume. The most-cited papers in this volume are "Instincts and their Vicissitudes," "Repression," "The Unconscious," "A Metapsychological Supplement to the Theory of Dreams," "Mourning and Melancholia," "A Case of Paranoia Running Counter to the Psycho-Analytic Theory of the Disease," "Thoughts for the Times on War and Death," and "On Transience," all written between 1915 and 1917. To simplify matters, we have listed the entire Volume 14 as Freud's most-cited work, even though he never published it as such.

Four hundred and thirty-three different authors wrote these 250 works! Eight authors are on the list both as primary and as co-authors: M Gell-Mann, C Djerassi, J Monod, L A Carlson, L D Landau, G E Palade, D H Spackman, and S Weinberg.

More than half (132) of the listed publications have only one author; 71, two authors; 29, three authors; 10, four authors; and 4, five authors. One paper has six, and two have seven authors.

During the past quarter-century the proportion of scientific papers having more than one author has increased significantly. Instead of working alone, as most did before World War II, many of our best scientists now work in teams.

As Derek J. De Solla Price of Yale University has reported, "Surprisingly enough, a detailed examination of the incidence of collaborative work in science shows that this is a phenomenon which has been increasing steadily and ever more rapidly since the beginning of the century.... Since that time the proportion of multi-author papers has acclerated steadily and powerfully, and it is now so large that if it continues at the present rate, by 1980 the single-author paper will be extinct."⁸

Price and others have suggested that the proportion of collaborative authorship in a field is related to the amount of economic support it receives. Price comments that "the amount of collaborative authorship measures no more than the economic value accorded to each field by society. A soft subject highly subsidized would become as collaborative as high energy physics...."9 A recent study by Henry J. Petroski of the Argonne National Laboratory supports Price's conclusions.10

Although there is general consensus that collaboration is increasing, there is little agreement on how collaboration affects citation analysis. Some researchers claim that the effects are negligible, while others state that they are "intolerable."11

In Social Stratification in Science, Jonathan and Stephen Cole studied

a wide range of citation data on 120 physicists. They found that, "The correlation between a straight citation count and total citations (including citations to collaborative work on which the physicist was not first author) is .96." The Coles suggest that "the omission of collaborative citations to papers on which the author was not first among collaborators does not affect substantive conclusions."12 But this depends upon what phenomenon you are studying.

In a recent study, Duncan Lindsey of Cornell University and George Warren Brown of Washington University reached an opposite conclusion. They argued that "one of the more serious errors in empirical judgment made in the field of the sociology of science has been to measure both publications and citations with counting procedures that do not take into account multiple authorship." They explore several alternatives to the first-author dilemma, including a variety of weighting schemes such as awarding the first author of a two-author paper two-thirds credit, and the second author, one-third. However, they conclude, "Until it becomes possible to decompose the relative contribution of collaborators, it will be necessary to simply divide by the number of contributors and allocate the credit equally."11

Lindsey and Brown also used the SCI and Social Sciences Citation Index $^{\text{TM}}$ (SSCI $^{\text{TM}}$) to determine the proportion of collaborative papers in a variety of fields. Their results indicate wide variations from one field to another. For biochemistry, they found that 19% of the 155 papers sampled had one author; 46%, two; 22%, three; and 13%, four or more. For psychology (205 papers), the breakdown was 75%, one author; 21%, two; 3%, three; and only 1%, four or more. In economics, 83% of the 154 papers sampled had one author; 16%, two; 1%, three; and none had four.

In the past, there were two primary reasons for ISI[®] to largely ignore the first-author problem in various citation analyses. First, many of the most-cited authors did their important work in the first half of this century, when collaboration was less pervasive than at present. Second, many authors who did publish as part of a team also published many papers alone. Their "wrapup" papers tended to be cited by others in the same way that review papers are now sometimes cited: as surrogates for groups of papers that characterize a particular research front.

The first-author "problem" in citation analysis is partly an artifact of the way SCI data is listed. To print the names of all authors of cited items would more than double the size of the SCI—without significantly increasing its value for information retrieval. But the data on coauthors is not lost, either in our printed indexes or on our tapes. We list only first authors in the citation indexes. But in the source indexes we include all co-authors of each item, as well as a full bibliographic description. Thus we have been able to use our own source data tapes from earlier years, in combination with citation data tapes, to compile all-author citation counts for over $4\frac{1}{2}$ million source articles indexed in the *SCI* from 1961 to 1976.

This new data, which we are just now beginning to study, classifies authors in six ways. It lists authors by overall rank, by primary-author rank, by secondary-author rank, and alphabetically. A "residue" author ranking includes citations to books and journal articles published before 1961. And an alphabetic listing is provided for the "residue" authors.

There are several important differences between our new all-author data and the data which we have used in the past. First, of course, is the fact that the all-author data credits citations equally to all coauthors of a given publication, not just the first author. However, it was not feasible for us to obtain and process the names of all co-authors of material published before 1961. Citations to this material are included in the "residue" listing. Thus, for example, since Lowry's classic paper was published in 1951. his name appears only in the "residue" listing. The same is true for many Nobel laureates and other eminent scientists whose significant work was done in the forties and the fifties

One advantage of the all-author data is that it will enable us to accourt for self-citations that previously were difficult to detect. Through a pattern of self-citation, one large research group could build up substantial citation counts based on its own local invisible college. It would be convenient to sociologists if such groups took on pseudonyms like the famous 'mathematician'' Bourbaki (actually a group), who appears in this list. But the in-breeding that takes place in science is often difficult to define. When can we say that a particular scientist has severed himself from his "family" and established a truly new group? And how can we determine the true extent to which particular scientists have made an impact on families of "offspring" scientists working both at the "birthplace" and elsewhere? The allauthor data can simplify finding answers to such questions.

What is the effect of using allauthor data to compile a list of mostcited authors? Since this list is based only on primary-author data, we might expect some significant changes when all-author data is used. In fact, when we compared these 250 authors with the 250 most-cited authors (overall) based on all-author data, only 69 names— 28% of the total—were the same.

To conclude this series of editorials, let me remind you that our list of 250 most-cited primary authors cannot be perfect or complete. The omission of one author's name has symbolized this lack of completeness. But what is most important about this list is that we have not had to be familiar with the work of all the authors involved in order to select them. We have not had to read the authors' works or consult with their peers. Yet we have been able to produce a list of individuals who comprise a distinguished elite among scientists.

I am well aware of the possible failure of citation analysis in some cases, particularly in those of lesscited individuals. It is much easier to quarrel with the "evaluation" of a poorly-cited individual than to deny the reality of high citation impact.

Of the one million or more authors who have published between 1961 and 1975, we have looked at the top two-and-a-half hundredths of one percent. Considering the elaborate apparatus, such as peer review, needed to make selections for grants, honors, and even employment and tenure, I think the method we have used to construct this list deserves further consideration and refinement.

Basically, I regard myself as an apolitical person. But life is politics in one form or another, and politics in science-as in other walks of life—has its own peculiar set of injustices. Is there any way to minimize these injustices? The mechanisms by which scientific recognition is achieved are certainly political. But so long as ability, insight, talent, and genius are unevenly disamong scientists, tributed we should try to insure that the political system of science which grants recognition is as fair and as democratic as possible. I believe that citation analysis can further that objective.

REFERENCES

- 1. Garfield E. The 250 most-cited primary authors. Part I. How the names were selected. Current Contents No. 49, 5 Dec 1977, p. 5-15.
- 3. Barrer R M. Diffusion in and through solids. New York: Macmillan, 1941. 464 pp.
- 4. Arnon D I. Personal communication. 30 November 1977.
- 5. Bridgman P W. The way things are. Cambridge: Harvard University Press, 1959. 333 pp.
- 7. Cochran W G. Citation Classics. Current Contents No. 19, 9 May 1977, p. 10.
- 8. Price D J D. Little science, big science. New York: Columbia University Press, 1963, 118 pp.
- Gitation measures of hard science, soft science, technology & nonscience. Communication Among Scientists and Engineers (Nelson C E & Pollock D K, eds.) Lexington, Mass.: Heath & Co., 1970, p. 7-22.
- Petroski H J. Trends in the applied mechanics literature. Technological Forecasting and Social Change 10:309-18, 1977.
- 11. Lindsey D & Brown G W. Problems of measurement in the sociology of science: taking account of collaboration. [Unpublished] 1977.
- 12. Cole J R & Cole S. Social stratification in science. Chicago: University of Chicago Press, 1973. 283 pp.

Figure 1. Each primary author's most-cited publication. Names of the 250 most-cited primary authors, 1961-1975, appear in boldface. The listed publication is the most-cited report of original research for which the author was primary author. Total citation count based on *Science Citation Index*[®] data.

Total Citations 1961-197		
1.	422	Abragam A. The principles of nuclear magnetism. New York: Oxford, 1961. 599 pp.
2.	5,241	Abramowitz M & Stegun I. Handbook of mathematical functions with formulas, graphs & mathematical tables. New York: Dover, 1964. 1046 pp.
3.	851	Abrikesov A A. On the magnetic properties of superconductors of the second type. Zh. Eksp. Teo. 32:1442-52, 1952. (Sov. Phys. JETP 5:1174-82, 1957.)
4.	201	Albert A, Goldacre R & Phillips J. The strength of heterocyclic bases. J. Chem. Soc. p. 2240-9, 1948.
5.	167	Allinger N L, Hirsch J A, Miller M A, Tyminski I J & Van-Catledge F A. Conformational analysis. Part 60. Improved calculations of the structures and energies of hydrocarbons by the Westheimer method. J. Am. Chem. S. 1199-1210, 1968.
6.	194	Allison A C, Denman A M & Barnes R D. Cooperating and controlling functions of thymus-derived lymphocytes in relation to autoimmunity. Lancet 2:135-40, 1971.
7.	409	Anden N E, Dahlstrom A, Fuxe K, Larsson K, Olson L & Ungerstedt U. Ascend- ing monoamine neurons to the telencephalon and diencephalon. <i>Acta. Physl. S.</i> 67:313-26, 1966.
8.	769	Anderson P W. Localized magnetic states in metals. Phys. Rev. 124:41-53, 1961.
9.	2,080	Andrews P. Estimation of the molecular weights of protein in Sephadexgel- filtration. <i>Biochem. J.</i> 91:222-33, 1964.
10.	1,806	Arnon D I. Copper enzymes in isolated chloroplasts; polyphenoloxidase in Beta Vulgaris. Plant Physl. 24:1-15, 1949.
11.	408	Axelrod J & Tomchick R. Enzymatic O-methylation of epinephrine and other catechols. J. Biol. Chem. 233:702-5, 1958.
12.	135	Baker B R. Design of active-site-directed irreversible enzyme inhibitors; the or-
13.	1,662	ganic chemistry of the enzymic active-size. New York: Wiley, 1967. 325 pp. Bardeen J, Cooper, L N & Scheiffer J R. Theory of superconductivity. Phys. Rev. 108:1175-1204, 1957.
14.	46	Barrer R M , Baynham J W, Bultitude F W & Meier W M. Hydrothermal chemis- try of the silicates. Part 8. Low temperature crystal growth of aluminosili-
		cates, and some gallium and germanium analogues. J. Chem. Soc. p. 195-208, 1959.
15.	184	
16.	267	Barton D H R. The stereochemistry of cyclohexane derivatives. J. Chem. Soc. p. 1027-40, 1953.
17.	88	Basolo F & Pearson R G. The transeffect in metal complexes. Prog. Inorg. Chem. 4:381-453, 1962.
18.	101	 Basov N G, Grasyuk A Z, Zubarey I G, Katulin V N & Krokhin O N. Semicon- ductor quantum generator with two photon optical excitation. Zh. Eksp. Teo. 50:551-9, 1966. (Sov. Phys. JETP 23:366-71, 1966.)
19.	416	Bates D R & Damgaard A. The calculation of the absolute strengths of spectral
20.	104	 lines. Phi. T. Roy. A. 242:101-22, 1949. Bell R P & Goodall D M. Kinetic hydrogen isotope effects in the ionization of some nitroparaffins. P. Roy. Soc. A. 294:273-97, 1966.
21.	7,300	Bellamy L J. The infra-red spectra of complex molecules. New York: Wiley, 1954. 323 pp.
22.	266	Bellman R E. Introduction to matrix analysis. New York: McGraw, 1960. 328 pp.
23.	451	Bender M L. Mechanism of catalysis of nucleophilic reactions of carboxylic acid derivatives. Chem. Rev. 60:53-113, 1960.

24. 228 Benson S W. Bond energies. J. Chem. Educ. 42:502-18, 1965.

Total Citations			
	1961-19		
25.	414	Bergstrom S, Carlson L A & Weeks J R. The prostaglandins: a family of bio- logically active lipids. <i>Pharm. Rev.</i> 20:1-48, 1968.	
26.	518	Berson S A , Yalow R S, Bauman M, Rothschild A & Newerly K. Insulin-1131 metabolism in human subjects: demonstration of insulin binding globulin in the	
27.	545	materie. (On the theory of the passage of corpuscular rays through matter).	
28.	356	Ann. Physik 5:325-400, 1930. Beutler E. The glutathione instability of drug-sensitive red cells. J. La. Cl. Med. 49:84-95, 1957	
29.	443		
30.	132	Birch A J. The reduction of organic compounds by metal-ammonia solutions. O. Rev. Chem. Soc. 4:69-93, 1950.	
31.	551	Bjorken J D. Asymptotic sum rules at infinite momentum. Phys. Rev. D. 79:1547-53, 1969.	
32.	1,145	Bloembergen N, Purcell E M & Pound R V. Relaxation effects in nuclear magnetic resonance absorption. <i>Phys. Rev.</i> 73:679-712, 1948. [Citation Classics. <i>Current Contents</i> [®] No. 18, 2 May 1977, p. 7.]	
33.	864	Born M & Huang K. Dynamical theory of crystal lattices. New York: Oxford, 1954. 420 pp.	
34.	476		
35.	1,203	Boyer P D. Spectrophotometric study of the reaction of protein sulfhydryl groups with organic mercurials. J. Am. Chem. S. 76:4331-7, 1954.	
36.	130	Brachet J , Denis H & DeVitry F. The effects of antimonycin D and puromycin on morphogenesis in amphibian eggs and Acetabularia mediterranea.	
37.	168	Develop. Bio. 9:398-434, 1964. Braunwald E & Frahm C J. Studies on Starling's law of the heart. Circulation 24:633-42, 1961.	
38.	6,952	Bray G A. A simple efficient liquid scintillator for counting aqueous solutions in a liquid scintillation counter. <i>Analyt. Bioc.</i> 1:279-85, 1960.	
39 .	126	[Citation Classics. Current Contents No. 2, 10 Jan 1977, p. 16.] Brldgman P W. The compression of 39 substances to 100,000 KG/CM. P. Am. Acad. Art Sci. 76:55-70, 1948.	
40.	328	Brodie B B, Gillette, J R & LaDu B N. Enzymatic metabolism of drugs and other foreign compounds. Ann. R. Bioch. 27:427-54, 1958.	
41.	544	Brown H C & Okamoto Y. Directive effects in aromatic substitution. Part 30. Electrophilic substituent constants. J. Am. Chem. S. 80:4979-87, 1958.	
42.	918	Brown J B. A chemical method for the determination of oestriol, oestrone, and oestradiol in human urine. <i>Biochem. J.</i> 60:185-93, 1955.	
43.	508	Buckingham A D. Chemical shifts in the magnetic resonance spectra of molecules containing polar groups. Can. J. Chem. 38:300-7, 1960.	
44.	275		
45.	315	Bunnett J F. Kinetics of reactions in moderately concentrated aqueous acids. I. Classification of reactions. J. Am. Chem. S. 83:4956-67, 1961.	
46.	811	Burn J H & Rand M J. The action of sympathomimetic amines in animals treated with reserpine. J. Physl. Lon. 144:314-36, 1958.	
47.	175	Burnet F M. The clonal selection theory of acquired immunity. Nashville: Vanderbilt University Press, 1959. 208 pp.	
48.	5,037	 Burton K. A study of the conditions and mechanism of the diphenylamine reaction for the colorimetric estimation of deoxyribonucleic acid. Biochem. J. 62:315-23, 1956. [Citation Classics. Current Contents No. 26, 27 Jun 1977, p. 23.] 	

•

Total Citations 1961-197		
49.	448	Busing W R, Martin K O & Levy H A. ORFLS, a FORTRAN crystallographic least-squares program. Report ORNL-TM-305. Oak Ridge, Tenn.: Oak Ridge National Laboratory, 1962. 77 pp.
50.	349	Carlson L A & Waldeck S. Determination of serum triglycerides. J. Atherosci. 3:334-6, 1963.
51.	506	Carlsson A. A fluorimetric method for the determination of dopamine (3-hydroxytyramine). Acta. Physl. S. 44:293-8, 1958.
52.	65	Cattell R B. The Scree test for the number of factors. Mult. Behav. Res. 1:245-76, 1966.
53.	870	Chance B & Williams G R. The respiratory chain and oxidative phosphorylation. Adv. Enzymol. 17:65-134, 1956.
54.	876	Chandrasekhar S. Stochastic problems in physics and astronomy. Rev. M. Phys. 15:1-89, 1943.
55. 56.	461 262	 Chapman S & Bartels J. Geomagnetism. Oxford: Clarendon, 1940. 2 vols. Chatt J & Duncanson L A. Olefin co-ordination compounds. Part 3. Infra-red spectra and structure: attempted preparation of acetylene complexes. J. Chem. Soc. p. 2939-47, 1953.
57.	948	Clementi E. Ab initio computations in atoms and molecules. IBM J. Res. 9:2-19, 1965.
58.	380	Cohen M H & Turnbull D. Molecular transport in liquids and glasses. J. Chem. Phys. 31:1164-9, 1959.
59.	1,529	Conney A H. Pharmacological implications of microsomal enzyme induction. Pharmacol. Rev. 19:317-66, 1967.
60.	108	Cope A C. Transannular reactions in medium sized rings. Q. Rev. Chem. S. 20:119-52, 1966.
61.	349	Corey E J. Dimethyloxosulfonium methylide [(CH ₃) SOCH ₂] and dimethyl- sulfonium methylide [(CH ₃) ₂ SCH ₂] formation and application to organic synthesis. J. Am. Chem. S. 87:1353-64, 1965.
62.	266	Cotton F A. Vibrational spectra and bonding in metal carbonyls. 3. Force constants and assignments of CO stretching modes of various molecules; evaluation of CO bond orders. <i>Inorg. Chem.</i> 3:702-11, 1964.
63.	207	Coulson C A & Longuet-Higgins H C. The electronic structure of conjugated systems, 1. General theory. P. Roy. Soc. Lond. 191:39-60, 1947.
64.	890	Courant R & Hilbert D. <i>Methods of mathematical physics.</i> New York: Wiley, 1959. 2 vols.
65.	248	 Cram D J & Elhafex F A A. Studies in stereochemistry. 10. The rule of "steric control of asymmetric induction" in the syntheses of acyclic systems. J. Am. Chem. S. 74:5828-35, 1952.
66.	1,780	Cromer D T & Waber J T. Scattering factors computed from relativistic Dirac- Slater wave functions. Act. Cryst. 18:104-9, 1965.
67.	394	Cruickshank D W J. The role of 3-d orbitals in π -bonds between (a) silicon, phosphorus, sulphur or chlorine and (b) oxygen or nitrogen. J. Chem. Soc. p. 5486-504, 1961.
68 .	709	Cuatrecases P. Protein purification by affinity chromatography; derivations of agarose and polyacrylamide beads. J. Biol. Chem. 245:3059-65, 1970.
69.	215	Curtis D R & Watkins J C. The excitation and depression of spinal neurones by structurally related amino acids. J. Neurochem. 6:117-41, 1960.
70.	121	Dacie J V , Grimes A J, Meisler A, Steingold L, Hemsted E H, Beaven G H & White J C. Hereditary heinzbody anaemia. <i>Br. J. Haem.</i> 10:388-402, 1964.
71.	282	Dalgarno A. Atomic polarizabilities and shielding factors. Adv. Phys. 11:281-315, 1962.
72.	6,342	
73.	596	
74.	1,402	DeDuve C. Tissue fractionation studies. 6. Intracellular distribution patterns of enzymes in rat liver tissue. <i>Biochem. J.</i> 60:604-17, 1955. [Citation Classics. <i>Current Contents</i> No. 12, 21 Mar 1977, p. 11.]

	litation	
1	961-197	75 Primary Author's Most-Cited Publication
75.	407	DeRobertis E, Pellegrino A D, Arnaiz G R & Salganicoff L. Cholinergic and non-cholinergic nerve endings in rat brain-1. J. Neurochem. 9:23-35, 1962.
76.	269	Dewar M J S & Schmeising H N. A re-evaluation of conjugation and hyper- conjugation: the effects of changes in hybridisation on carbon bonds. <i>Tetrahedron</i> 5:166-78, 1959.
	1,326	Dische Z. A specific new color reaction by hexuronic acids. J. Biol. Chem. 167:189-98, 1947.
	1,097	Dixon M. The determination of enzyme inhibitor constants. Biochem. J. 55:170-1, 1953.
79. 90	440	Djerassi C , Engle R R & Bowers A. The direct conversion of steroidal \triangle^5 . 3 β -alcohols to \triangle^5 - and \triangle^4 -3-ketones. J. Org. Chem. 21:1547-9, 1956.
80.	273	Doering W V & Hoffman A K. The addition of dichlorocarbene to olefins. J. Am. Chem. S. 76:6162-5, 1954.
	2,705	 Dole V P. A relation between non-esterified fatty acids in plasma and the metabolism of glucose. J. Clin. Inv. 35:150-4, 1956. Duncan D B. Multiple range and multiple F tests. Biometrics 11:1-42, 1955.
		[Citation Classics. Current Contents No. 4, 24 Jan 1977, p. 10.] Eagle H. Amino acid metabolism in mammalian cell cultures.
65.	2,355	Science 130:432-7, 1959. [Citation Classics. Current Contents No. 5, 31 Jan 1977, p. 13.]
84.	345	Eccles J C, Fatt P & Koketso K. Cholinergic and inhibitory synapses in a pathway from motor axon collaterals to motoneurons.
85.	421	J. Physl. Lon. 126:524-62, 1954. Eigen M. Protonenübertragung, säure-base katalyse und enzymatische
		hydrolyse. I. Elementarvorgänge'. Angew. Chem. 75:489-515, 1963. (Proton- transfer, acid-base catalysis and enzymatic hydrolysis. I. Elementary processes. Angew. Chem. Int. 3:1-19, 1964.
86.		J. Am. Chem. S. 90:3444-58, 1968.
87.	213	Erdelyl A. Higher transcendental functions. Vol. 1. New York: McGraw-Hill, 1953. 302 pp.
88.	177	Eysenck H J. The biological basis of personality. Springfield, Ill.: Thomas, 1967. 399 pp.
89. 00	965	ulins in antibody-agar plates. J. Immunol. 94:84-90, 1965.
9 0.	1,015	and related compounds condensed with formaldehyde. J. Hist. Cyto. 10:348-54, 1962.
91.	1,086	Farquhar M G & Palade G E. Junctional complexes in various epithelia. J. Cell Biol. 17:375-412, 1963.
92.	283	J. Hist. Cyto. 13:75-91, 1965.
	1,111	Feigl F. Spot tests in organic analysis. New York: Elsevier, 1956. 2 vols.
94.	263	Feldberg W & Lewis G P. The action of peptides on the adrenal medulla. Release of adrenaline by bradykinin and angiotensin. J. Physl. Lon. 171:98-108, 1964.
95.	770	Feynman R P & Gell-Mann M. Theory of the Fermi interaction. Phys. Rev. 109:193-8, 1958.
96.	133	Fleser L F. Preparation of ethylenethioketals. J. Am. Chem. S. 76:1945-7, 1954
97.	116	Fischer E O & Maasbol A. Phenyl-methoxycarben-und methylmethoxycarben- pentacarconylchrom,-molybdan, -wolfram und -cyclopentaienyl-dicarbonyl- mangan. (Pentacarbonylchromium, -molybdenum, -tungsten and cyclopenta- dienyl-dicarbonylmanganese complexes of phenylmethoxycarbene and me- thylmethoxycarbene. Chem. Ber. 100:2445-56, 1967.
98.	623	Fischer M E. The theory of equilibrium critical phenomena. Rep. Pr. Phys. 30:615-730, 1967.

C	otal litation: 961-197	
100.	7,395	Fiske CH & Subbarow Y. The colorimetric determination of phosphorus. J. Biol. Chem. 66:375-400, 1925.
101.	407	
		Ithaca, N.Y.: Cornell University Press, 1953. 672 pp.
102.	7,454	Folch J, Lees, M & Sloane-Stanley G H. A simple method for the isolation and purification of total lipides from animal tissues. J. Biol. Chem. 226:497-509, 1957.
103.	805	Fraenkel-Conrat H, Harris J I & Levy A L. Recent developments in techniques for terminal and sequence studies in peptides and proteins. <i>Methods of bio- chemical analysis.</i> (Glick D. ed.) New York: Wiley-Interscience, 1955. Vol. 2, pp. 359-425.
104.	3,638	
		integrated approach to mechanics and disorders.
		N. Eng. J. Med. 276:34-44, 94-103, 148-56, 215-25, 273-81, 1967.
105.	116	Freud S. On the history of the psychoanalytic movement; papers on meta-
		psychology; and other works. Vol. 14 of Standard edition of the complete
		psychological works of Sigmund Freud. (Strachey J & Freud A. eds.)
106.	324	London: Hogarth Press, 1957. 374 pp. Friedel J. Metallic alloys. Nuov. Ciment. 7:287-311, 1958.
	1,397	Gell-Mann M. Symmetries of baryons and mesons. Phys. Rev. 125:1067-84, 1962.
108.	224	Gilman H & Schulze F. A qualitative color test for the Grignard reagent.
		J. Am. Chem. S. 47:2002-5, 1925.
109.	407	Ginzburg V L & Landau L D. On the theory of superconductivity. Zh. Eksp. Teo. 30:1064-82, 1950.
110.	833	Glasstone S, Laidler K J & Eyring H. Theory of rate processes. New York: McGraw-Hill, 1941. 611 pp.
111.	514	Gomori G. A modification of colorimetric phosphorus determination for use with the photoelectric colorimeter. J. La. Cl. Med. 27:955-60, 1942.
112.	184	Good R A, Kelly W O, Rotstein J & Varco R L. Immunological deficiency diseases. <i>Prog. Allerg.</i> 6:187-319, 1962.
113.	436	Goodman L S & Gilman A. Pharmacological basis of therapeutics. New York: Macmillan, 1941. 1383 pp.
114.	884	Goodwin T W. The spectrophotometric determination of tyrosine and tryptophan in proteins. <i>Biochem. J.</i> 40:628-32, 1946.
115.	5,396	Gornall A G, Bardawill C J & Daird M M. Determination of serum proteins by means of the biuret reaction. J. Biol. Chem. 78:751-66, 1949.
116.	959	Grabar P & Williams C A. Métnode permettant l'étude conjuguée des propriétés électrophorétiques et immunochimiques d'un mélange de protéines. Appli- cation au sérum sanguin. (Method of combined study of electrophoretic and immunochemical properties of a protein mixture. Application to blood serum.) Bioc. Biop. A. 10:193-4, 1953.
117.	145	Granit R A & Kaada B R. Influences of stimulation of central nervous structures in cat. Act. Physl. S. 27:130-60, 1953.
118.	250	Green D E & Fleisher S. The role of lipids in mitochondrial electron transfer and oxidative phosphorylation. <i>Bioc. Biop. A.</i> 70:554-82, 1963.
119.	652	 Gutowsky H S & Holm C H. Rate processes and nuclear magnetic resonance spectra. 2. Hindered internal rotation of amides. J. Chem. Phys. 25:1228-34, 1956.
120.		Hansen M. Constitution of binary alloys. New York: McGraw-Hill, 1958. 1305 pp.
121.	1,841	Harned H S & Owen B B. The physical chemistry of electrolytic solutions.
122.	659	New York: Reinhold, 1943. 611 pp. Herbert V, Lau K S, Gottlieb C W & Bleicher S J. Coated charcoal immuno-
122.	479	assay of insulin. J. Clin. End. 25:1375-84, 1965. Herzberg G. Molecular spectra and molecular structure. Vol. 1: Spectra of
	479	diatomic molecules. New York: Van Nostrand, 1950. 616 pp. Hirs C H W. The oxidation of ribonuclease with performic acid.
124.	1,193	J. Biol. Chem. 219:611-21, 1956.

C	fotal Sitation 961-197	
		Hirschfelder J O, Curtis C F & Bird R B. Molecular theory of gases and
	-	liquids. New York: Wiley, 1954. 1219 pp.
126.	1,089	Hodgkin A L & Huxley A F. A quantitative description of membrane current and its application to conduction and excitation in nerve. J. Physl. Lon. 117:500-44, 1952.
127.	92	Horner L & Winkelmann E H. Neuere methoden der praparativen organischen chemie. 2. (Recently developed methods for preparatory organic chemistry. 2 Angew. Chem. 71:349-65, 1959.
128.	192	 House H O, Respess W L & Whitesides G M. The chemistry of carbanions. 12 The role of copper in the conjugate addition of organometallic reagents. J. Org. Chem. 31:3128-41, 1966.
129.	916	Hubel D H & Wiesel T N. Receptive fields, binocular interaction and func- tional architecture in the cat's visual cortex. J. Physl. Lon. 160:106-54, 1952
130.	753	Hulsgen R. 1.3-dipolarecycloadditionen. Rückshau und ausblick. Angew. Chem. 75:604-37, 1963. (1,3-dipolar cycloadditions; past and future.
	(50	Angew. Chem. Inter. 2:565-98, 1963.)
131.	2,011	Huxley H E. Electron microscope studies on the structure of natural and synthetic protein filaments from striated muscle. J. Mol. Biol. 7:281-308, 1963 Ingold C K. Structure and mechanism in organic chemistry.
132.	2,011	Ithaca, N.Y.: Cornell University Press, 1953. 282 pp.
133.	2,071	Jackman L M & Sternhell S. Applications of nuclear magnetic resonance spectroscopy in organic chemistry.
174	2 227	Elmsford, N.Y.: Pergamon, 1959. 134 pp.
134.	2,227	 Jacob F & Monod J. Genetic regulatory mechanisms in the synthesis of proteins. J. Mol. Biol. 3:318-56, 1961. [Citation Classics. Current Contents No. 33, 15 Aug 1977, p. 9.]
135.	1,667	Jaffé H H. A re-examination of the Hammett equation. Chem. Rev. 53:191-261, 1953.
136.	387	Johnson H L & Morgan W W. Fundamental stellar photometry for standards of spectral type on the revised system of the Yerkes spectral Atlas.
137.	196	 Astrophys. J. 117:313-52, 1953. Jorgensen C K. Comparative crystal field studies of some ligands and the lowest state of paramagnetic nickel (II) complexes. Act. Chem. S. 9:1362-77, 1955.
138.	3,240	Kabat E A & Mayer M. Experimental immunochemistry. Springfield, Ill.: C C Thomas, 1948. 905 pp.
	1,689	Karnovsky M J. A formaldehyde glutaraldehyde fixative of high osmolality for use in electron microscopy. J. Cell Bio. 27:A137-8, 1965.
	1,629	Karplus M. Contact electron spin coupling of nuclear magnetic moments. J. Chem. Phys. 30:11-5, 1959.
141. 142.	759	 Kato T. Perturbation theory for linear operators. New York: Springer, 1966. 592 pp. Katritzky A R. The infrared spectra of heteroaromatic compounds.
142.		Q. Rev. Chem. Soc. 13:353-73, 1959. Katz B. The transmission of impulses from nerve to muscle, and the sub-
143.		 cellular unit of synaptic section. P. Roy. Soc. B. 155:455-77, 1962. Kellin D & Hartree E F. On the mechanism of the decomposition of hydrogen
	100	peroxide by catalese. P. Roy. Soc. B. 124:397-405, 1938.
145.	443	Kety S S & Schmidt C F. The effects of altered arterial tensions of carbon dioxide and oxygen on cerebral oxygen consumption of normal young men.
146.	39 7	J. Clin. Inv. 27:484-92, 1948. King R B. Organometallic syntheses. Vol. 1. Transition-metal compounds. New York: Academic Press, 1965. 186 pp.
147.	277	Kirkwood J G & Riseman J. The intrinsic viscosities and diffusion con-
148.	253	stants of flexible macromolecules in solution. J. Chem. Phys. 16:565-73, 1948 Kittel C. Physical theory of ferromagnetic domains.

Rev. M. Phys. 21:541-83, 1949.

Total Citations 1961-197	
149. 384	Klein G, Sjogren H O, Kelin E & Hellstrom K E. Demonstration of resistance against methylcholanthrene-induced sarcomas in the primary auto- chthonous host. <i>Cancer Res.</i> 20:1561-72, 1960.
150. 216	Klotz I M, Walker M & Pivan R B. The binding of organic ions by pro- teins. J. Am. Chem. S. 68:1486-90, 1946.
151. 147	Kolthoff I M, Bruckenstein S & Chantooni K M. Acid base equilibria in acetonitrile. Spectrophotometric and conductometric determination of the dissociation of various acids. J. Am. Chem. S. 83:3927-35, 1961.
152. 162	Kornberg A & Pricer W E. Enzymatic esterification of 2-glycerophosphate by long chain fatty acids. J. Biol. Chem. 204:345-57, 1953.
153. 1,535	Krebs H A & Henseleit K. Untersuchungen über die Harnstoffbildung im Tierkörper. (Studies on urea formation in the animal organism.) H-S Z. Physl. 210:33-66, 1932.
15 4 . 947	
155. 412	Kuhn R, Trischmann H & Low I. Zur permethylierung von zuckern und glykosiden. (Permethylation of sugars and glycosides.) Angew. Chem. 67:32, 1955.
156. 214	Landau L D. The theory of superfluidity of helium II. J. Phys. USSR. 5:71-90, 1941.
157. 268	Lee T D, Weinberg S & Zumino B. Algebra of fields. Phys. Rev. L. 18:1029-32, 1967.
158. 412	Lehninger A L. Water uptake and extrusion by mitochondria in relation to oxidative phosphorylation. <i>Physl. Rev.</i> 42:467-517, 1962.
159. 605	Lemieux R U, Kullnig R K, Bernstein H J & Schneider W G. Configurational effects on the proton magnetic resonance spectra of six membered ring compounds. J. Am. Chem. S. 80:6098-6105, 1958.
160. 101	Levine S & Mollins R F. Hormonal influences on brain organization in infant rats. Science 152:1585-92, 1966.
161. 4,633	Lineweaver H & Burk D. The determination of enzyme dissociation constants. J. Am. Chem. S. 56:658-66, 1934.
162. 453	
163.50,016	Lowry O H, Rosebrough N J, Faff A L & Randall R J. Protein measurement with the folin phenol reagent. J. Biol. Chem. 193:256-65, 1951. [Citation Classics. Current Contents No. 1, 3 Jan 1977, p. 7.]
164. 6,953	Luft J H. Improvements in epoxy resin embedding methods. J. Biop. Bioc. 9:409-14, 1961. [Citation Classics. Current Contents No. 20, 16 May 1977, p. 8.]
165. 3,148	
166. 518	McConnell H M & Chestnut D B. Theory of isotropic hyperfine interactions in π electron radicals. J. Chem. Phys. 28:107-17, 1958.
167 <i>.</i> 887	McKusick V A. Heritable disorders of connective tissue. St. Louis: Mosby, 1956. 224 pp.
168. 669	
169. 1,441	Millong G. Advantages of a phosphate buffer for OsO4 solutions in fixation. J. Appl. Phys. 32:1637, 1961.
170. 505	
171. 1,685	Monod J. On the nature of allosteric transitions: a plausible model. J. Mol. Biol. 12:88-118, 1965.
172. 1,475	Moore S, Spackman D H & Stein W H. Chromatography of amino acids on sulfonated polystyrene resins. <i>Analyt. Chem.</i> 30:1185-90, 1958.

Total Citation 1961-192		
173.	276	Morse P M. Diatomic molecules according to the wave mechanics. 2. Vibrational levels. Phys. Rev. 34:57-64, 1929.
174.	412	Mott N F. Electrons in disordered structures. Adv. Phys. 16:49-144, 1967.
175.	104	Muller A & Krebs B. Normal coordinate treatment of XY4- type molecules and ions with Td symmetry. J. Mol. Spect, 24:180-97, 1967.
176.	61	Müller E & Rundel W. Verätherung von alkoholen mit diazomethan unter borfluorid-katalyse. (Etherification of alcohols with diazomethane under boron fluoride catalysis.) Angew. Chem. 70:105, 1958.
177.	960	Mulliken R S. Electronic population analysis on LCAO-MO molecular wave functions. 1. J. Chem. Phys. 23:1833-40, 1955.
178.	240	Nakamoto K, Margoshes M & Rundle R E. Stretching frequencies as a function of distances in hydrogen bonds. J. Am. Chem. S. 77:6480-6, 1955.
179.	102	Natta G. Progress in the stereospecific polymerization. Makrom. Chem. 35:94-131, 1960.
180.	281	Nesmeyanov A N. Davlenie para khimicheskikh elementov. (Vapour pressure of the elements.) Moskow: Publishing House of the Academy of Sciences of USSR, 1961. 469 pp.
181.		Newman M S & Beal P F. An improved Wolff rearrangement in homogeneous medium. J. Am. Chem. S. 72:5163-5, 1950.
182.	390	Novikoff A B, Essner E & Quintana N. Golgi apparatus and lysosomes. Fed. Proc. 23:1010-22, 1964.
183.	123	Olah G A & White A M. Stable carbonium ions. 91. Carbon-13 nuclear magnetic resonance spectroscopic study of carbonium ions. J. Am. Chem. S. 91:5801-10, 1969.
184.	875	Ouchterlony O. Diffusion-in-gel methods for immunological analysis. Prog. Allergy 5:1-78, 1958.
185.	2,072	Palade G E. A study of fixation for electron microscopy. J. Exp. Med. 95:285-97, 1952.
186.	3,040	Pauling L. The nature of the chemical bond and the structure of molecules and crystals; an introduction to modern structural chemistry. Ithaca, N.Y.: Cornell University Press, 1939, 429 pp.
187.	1,256	Pearse A G E. Histochemistry, theoretical and applied. Boston: Little, Brown, 1953. 530 pp.
188.	471	Perutz M F, Muirhead H, Cox J M & Goaman L C G. Three dimensional Fourier synthesis of horse oxyhaemoglobin at 2.8 Å resolution: The atomic model, <i>Nature</i> 219:131-9, 1968.
	1,215	CNDO results for AB2 and AB3 systems. J. Chem. Phys. 44:3289-96, 1966.
190.	624	New York: Interscience, 1957. 448 pp.
191.		Racker E. Spectrophotometric measurements of the enzymatic formation of fumaric and cisacontic acids. <i>Bioc. Biop. A.</i> 4:211-4, 1950.
	3,323	Reed L J & Meunch H. A simple method of estimating fifty percent endpoints. Am. J. Hyg. 27:493-7, 1938. Percender F S. The use of lead citrate at high pH as an electron oneque stain
193. 194.	9,002 206	in electron microscopy. J. Cell Biol. 17:208-12, 1963.
174.	200	reactions of cyclobutyl, cyclopropylcarbinyl and allycarbinyl derivatives. J. Am. Chem. S. 73:2509-20, 1951.
195.	438	Robinson R A & Stokes R H. Electrolyte solutions. New York: Plenum, 1955. 559 pp.
196.	400	Rose M E. The analysis of angular correction and angular distribution data. <i>Phys. Rev.</i> 91:610-5, 1953.
197.	520	Rossini F D. Selected values of physical and thermodynamic properties of hydrocarbons and related compounds. Pittsburgh: Carnegie University Press, 1953. 1050 pp.
198.	307	Russell G A , Jazen E G & Strom E I. Electron transfer processes. 1. The scope of the reaction between carbanions or nitranions and unsaturated electron acceptors. J. Am. Chem. S. 86:1807-14, 1964.

Totai Citation 1961-197	
199. 4,934	Sabatini D D. Cytochemistry and electron microscopy: the preservation of cellu- lar ultrastructure and enzymatic activity by aldehyde fixation. J. Cell Biol. 17:19-58, 1963.
200. 1,575	Scatchard G. The attractions of proteins for small molecules and ions. Ann. N.Y. Acad. 51:660-72, 1949.
201. 3,660	
202. 1,518	 Schneider W C. Phosphorus compounds in animal tissues. 1. Extraction and estimation of desoxypentose nucleic acid and of pentose nucleic acid. J. Biol. Chem. 161:293-303, 1945. [Citation Classics. Current Contents No. 8, 21 Feb 1977, p. 12.]
203. 251	untersuchung von austausch-gleichgewichten. (Polarographic study on ex- change equilibria.) Helv. Chim A. 37:937-57, 1954.
204. 245	Schwinger J. Field theory commutators. Phys. Rev. L. 3:296-7, 1959.
205. 259	Seeger A. The mechanism of glide and work hardening in face-centered cubic and hexagonal close-packed metals. <i>Dislocations and mechanical properties</i> of crystals. (Fisher T C, Johnston W G, Thomson R & Vreeland T, eds.) New York: Wiley, 1957, pp. 243-329.
206. 249	Seltz F. Color centers in alkali halide crystals. 2. Rev. M. Phys. 26:7-94, 1954.
207. 167	Selye H. The general adaptation syndrome and the diseases of adaptation. J. Clin. End. 6:117-230, 1946. [Citation Classics. Current Contents No. 13, 28 Mar 1977, p. 13.]
208. 168	Seyferth D, Burlitch J M, Minasz R J, Mui J Y P, Simmons H D, Treiber A D H & Dowd S R. Halomethyl-metal compounds. 2. The preparation of gem-dihalo- cyclopropanes by the reaction of phenyl(trihalomethyl)mercury compounds with olefins. J. Am. Chem. S. 87:4259-70, 1950.
209. 167	Sillen L G. High speed computers as a supplement to graphic methods. Act. Chem. S. 16:159-72, 1962.
210. 752	Skou J C. The influence of some cations on an adenosine triphosphatase from peripheral nerves. <i>Bioc. Biop. A.</i> 23:394-401, 1957.
211. 772	Slater J C. Atomic shielding constants. Phys. Rev. 36:57-64, 1930.
212. 782	Smith H W, Finklestein N, Aliminosa L, Crawford B & Grabar M. The renal clearances of substituted hippuric acid derivatives and other aromatic acids in dog and man. J. Clin. Inv. 24:388-404, 1945.
213. 2,041	Smithles 0. Zone electrophoresis in starch gels: group variations in the serum proteins of normal human adults. <i>Biochem. J.</i> 61:629-41, 1955.
214. 3,254	Snedecor G W & Cochran W G. Statistical methods applied to experiments in agriculture and biology. Ames, Iowa: Iowa State, 1956. 534 pp. [Citation Classics. Current Contents No. 19, 9 May 1977, p. 10.]
215. 1,439	Somogyl M. Notes on sugar determination. J. Biol. Chem. 195:19-23, 1952.
216. 5,727 217. 276	 Spackman D H, Stein W H & Moore S. Automatic recording apparatus for use in the chromatography of amino acids. Analyt. Chem. 30:1190-1206, 1958. Spitzer L. Physics of fully ionized gases.
	New York: Interscience, 1956. 105 pp.
218. 382	Stahl E. Dünnschicht-chromatographie. II. Standardisierung, sichtbarmachung, dokumentation, und anwendung. (Thin layer chromatography. 2. Standardiza- tion, visualization, documentation and application.) <i>Chem. Zeitun.</i> 82:323-9, 1958.
219. 1,381	reference to the biological sciences. New York: McGraw-Hill, 1960. 481 pp. [Citation Classics. Current Contents No. 39, 26 Sep 1977, p. 20.]
220. 2,155	New York: Wiley, 1961. 489 pp.
221. 330	cemic-glycogenolytic factor from insulin and from gastric mucosa. J. Biol. Chem. 180:825-37, 1949.
222. 331	Taft R W. Sigma values from reactives. J. Phys. Chem. 64:1805-15, 1960.

0	Fotal Citation 1961-197	
223.	356	Tanford C. Protein denaturation. Adv. Protein Chem. 23:121-282, 1968.
224.	246	Udenfriend S, Weissbach H & Clark C T. The estimation of 5-hydroxytryptamin (serotonia) in biological tissue. J. Biol. Chem. 215:337-44, 1955.
225.	916	Umbreit W W, Burris R H & Stauffer J F. Manometric and biochemical techniques. Minneapolis: Burgess, 1945. 198 pp.
	1,176	 Van Slyke D D & McNeill J M. The determination of gases in blood and other solutions by vacuum extractions and manometric measurements. I. J. Biol. Chem. 61:523-73, 1924.
227.	907	Van Vleck J H. The dipolar broadening of magnetic resonance in crystals. Phys. Rev. 57:426-47, 1940.
228.	429	von Euler U S & Lishajko F. Improved technique for the fluorimetric estimation of catecholamines. Act. Phys. S. 51:348-56, 1961.
229.	132	 Walling C & Thaler W. Positive halogen compounds. 3. Allylic chlorination with tert-butyl hypochlorite stereochemistry of allylic radicals. J. Am. Chem. S. 83:3877, 1961.
	2,594	Warburg O. Isolierung und kristallisation des gärungsferments enolase. (Isolation and crystallization of the enzyme enolase.) Biochem. Z. 310:384-421, 1941.
	2,656	Warren L. The thiobarbituric acid assay of sialic acids. J. Biol. Chem. 234:1971-5, 1959.
	2,509	Watson M L. Staining of tissue sections for electron microscopy with heavy metals. J. Biop. Bioc. 4:475-9, 1958.
233.	242	Weber G. Polarization of the fluorescence of macromolecules. 2. Fluorescent conjugates of ovalbumin and bovine serum albumin. Biochem. J. 51:155-67, 1952.
234.	4,586	 Weber K & Osborn M. The reliability of molecular weight determinations by dodecyl sulfate-polyacrylamide gel electrophoresis. J. Biol. Chem. 244:4406-12, 1969.
235.	622	Weinberg S. Pion scattering lengths. Phys. Rev. L. 17:616-21, 1966.
236.	275	Weiss P. Experiments on the mechanism of nerve growth. J. Exp. Zool. 107:315-95, 1948.
237.	335	Wiberg K B. The deuterium isotope effect. Chem. Rev. 55:713-43, 1955.
238.	108	Wieland T. Poisonous principles of mushrooms of the genus amanita. Science 195:946-52, 1968.
239.	101	Wigglesworth V B. The function of the corpus allatum in the growth and re- production of rhodnius proplexus (Hemiptera). Q.J. Mic. Sci. 79:91-121, 1936.
240.	386	Wigner E P. On unity representations of the inhomogenous Lorentz group. Ann. Math. 40:149-204, 1939.
241.	503 1,444	Wilson E B. Molecular vibrations: theory of infra-red and raman vibrational spectra. New York: McGraw-Hill, 1955. 388 pp.
242.	1,444	Winer B J. Statistical principles in experimental design. New York: McGraw-Hill, 1962. 672 pp.
243.	366	 Winstein S & Holness N J. Neighboring carbon and hydrogen. 19. T-butylcyclo- hexl derivatives. Quantitative conformational analysis. J. Am. Chem. S. 77:5562-78, 1955.
244.	184	Wittig G & Schollkopf U. Uber triphenyl-phosphinmethylene als olefinbindende reagenzien. (Triphenyl-phosphinemethylenes as olefin-binding reagents.) Chem. Ber. 87:1318-30, 1954.
245.	962	Woodward R B, & Hoffmann R. Die erhaltung der orbitalsymmetric. (The conservation of orbital symmetry.) Angew. Chem. 81:797-869, 1969. (Angew. Chem. Int, 8:781-853, 1969.)
246.	419	Zacharlasen W H. The secondary extinction correction. Act. Cryst. 16:1139-44, 1963.
247.	561	Zeldovich Y B & Raizer Y P. Physics of shock waves and high temperature hydrodynamic phenomena. New York: Academic Press, 1966. 2 vols.
248.	316	Ziman J M. A theory of the electrical properties of liquid metals. 1. The monovalent metals. <i>Philosoph. Mag.</i> 6:1013-34, 1961.
249.	151	Zimmerman H E & Schuster D I. A new approach to mechanistic organic photochemistry. 4. Photochemical rearrangements of 4,4-diphenylcyclohexa- dienone. J. Am. Chem. S. 84:4527-40, 1962.